Astronomy 330

This class (Lecture 11):

Life in the Solar System

Life on Earth Scott Huber Thomas Hymel

Next Class:

Dave Dreiser Feifei Lian



HW 2

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- Kevin Homann
 <u>http://educate-yourself.org/dc/</u>
 orgonegenindex.shtml
- Michael Hutchinson
 http://www.abduct.com/

Presentations

Music: Jesus Came From Outta Space- Supergrass

- Scott Huber Element 115: Antigravity?
- Thomas Hymel The Challenges of a Generation

Outline

- n_e
- Why is Earth a good place for life?
- What about Venus or Mars?



Complex term, so let's break it into two terms:

- n_n: number of planets suitable for life per planetary system
- f_s: fraction of stars whose properties are suitable for life to develop

http://nike.cecs.csulb.edu/~kjlivio/Wallpapers/Planets%2001.jpg



Moon: Impact Implications



- Even if the moon theory is • Hot, Hot, Hot! incorrect, other smaller bodies were playing havoc on the surface.
- When they impact, they release kinetic energy and gravitational potential.
- In addition, some of the decaying radioactive elements heated up the Earth-stored supernova energy!
- The planetesimals melt, and the Earth went through a period of differentiation.



Planetary Differentiation



Differentiation: Iron Catastrophe

- Average density of Earth is 5.5 g/cm³
- Average density on the surface is 3 g/cm³
- So, something heavy must be inside
- When the Earth formed it was molten
 - Heavy materials (e.g. iron, nickel, gold) sank
 - Lighter materials (e.g. silicon, oxygen) floated to the top



Early Earth



http://www.black-cat-studios.com/catalog/earth.htm

Question



Which of the following does NOT well describe the early Earth?

- a) So hot that the surface had molten rock.
- b) There was no water.
- c) The surface kept getting hit by really, really big rocks.
- d) The oxygen rich atmosphere caused quick oxidation (rusting) of iron-rich rocks
- e) No chance of life at this stage.

Structure

• Luckily, not all of the iron sank to the center, else we would be still in the Stone Age.

• No atmosphere

• No water

• High temp

• No life.....

• Big rocks keep

falling on my head...

- Temperature increases as you go deeper underground. From around 290 K on surface to nearly 5000 K at center.
 - Heated by radioactive decay
 - Supernovae remnants
- Earth's magnetic field is established early on.. after the iron catastrophe... good for life.



The Crust

- Outside layer of the Earth (includes oceans) floats on top of still hot interior
 - About 50 km thick
 - Coldest layer rocks are rigid
- Mostly silicate rocks
 - Made of lighter elements like silicon, oxygen, and aluminum
- Oxygen and water are abundant
- Excellent insulator
 - Keeps the Earth's geothermal heat inside!



Today's Earth Surface



- 70% of the Earth's surface is covered with water
 - Ocean basins
 - Sea floors are young, none more than 200 million years old
- 30% is dry land Continents
 - Mixture of young rocks and old rocks
 - Up to 4.2 billion years old



Geologically Active Surface

- The young rocks on the Earth's surface indicate it is geologically active
- Where do these rocks come from?
 - Volcanoes
 - Rift valleys
 - Oceanic ridges
- Air, water erode rocks
- The surface is constantly changing





Recycling Bio-elements

- From gravity and radioactivity, the core stays hot.
- This allows a persisting circulation of bioelements through continental drift— melting of the crust and re-release through volcanoes.
- Otherwise, certain elements might get locked into sediment layers- e.g. early sea life.
- Maybe planets being formed now, with less supernovae, would not have enough radioactivity to support continental drifts and volcanoes. (Idea of Peter Ward and Donald Brownlee.)



http://www.pahala-hawaii.com/j-page/image/activevolcanoe.jpg



The Earth's 1st Atmosphere

- The inner disk had most gases blown away and the proto-Earth was not massive enough to capture these gases.
- Any impacts (e.g. the moon), would have blown any residual atmosphere away.
- The first atmosphere was probably H and He, which was lost quickly.



http://www.udel.edu/Biology/Wags/wagart/worldspage/impact.git

The Earth's 1st Atmosphere

- The interior heat of the Earth helped with the Earth's early atmosphere.
- Volcanoes released gases (water vapor and CO₂)
- Another scenario is that impacted comets released – water (H₂O), carbon dioxide (CO₂), and Nitrogen (N₂)
 – the first true atmosphere.
- The water condensed to form the oceans and much of the CO₂ was dissolved in the oceans and incorporated into sediments- such as calcium carbonate (CaCO₃).



http://www.fli-cam.com/images/comet-liner.jpg

Our Atmosphere

- Rocks with ages greater than <u>2 billion</u> years show that there was little or no oxygen in the Earth's atmosphere.
- The current composition: 78% nitrogen, 21% oxygen, and trace amounts of water, carbon dioxide, etc.
- Where did the oxygen come from?
- Cyanobacteria made it.
 - Life on Earth modifies the Earth's atmosphere.



http://www.uweb.ucsb.edu/~rixfury/conclusion.htm

This New Planet

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- Mostly oceans and some solid land (all volcanic).
- Frequent impacts of remaining planetesimals (ending about 3.8 billion years ago).
- Impacts would have sterilized the young Earth– Mass extinctions and



maybe vaporized any oceans (more comets?).

This New Planet

- Impacts and volcanic activity created the continental landmasses.
- Little oxygen means no ozone layer– flooded with ultraviolet light on surface.
- Along with lightning, radioactivity, and geothermal heat, provided energy for chemical reactions.
- BUT, life on the surface not possible!



Question



The Earth's first atmosphere was

- a) much like today's atmosphere, but older.
- b) Trick Question. There was no atmosphere.
- c) likely just H and He, and blown away quickly.
- d) made from comets.
- e) a combination of volcano gases and comet collisions.

CHOLO CALL

Water



- Water is a key to life on Earth.
- Primary constituent of life- "Ugly bags of mostly water"
 - Life is about 90% water by mass.
- Primary role as a solvent
 - Dissolves molecules to bring nutrients and remove wastes. Allows molecules to "move" freely in solution.
 - Must be in liquid form, requiring adequate pressure and certain range of temperatures.
- This sets a requirement on planets, if we assume that all life requires water.

Water as a Solvent

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- The water molecule is "polar". The oxygen atoms have more build-up of negative charge than the hydrogen. This allows water molecules to link up, attracted to each other.
- In this way, water attracts other molecules, surrounds them and effectively dissolves them into solution.





Example: Dissolving Table Salt

The partial charges of the water molecule are attracted to the Na⁺ and Cl ions. The water molecules work their way into the crystal structure and between the individual ions, surrounding them and slowly dissolving the salt.



Keeping it Useful: Atmosphere

- Need to have enough pressure to keep water from boiling away at low temperature
 - Cooking at higher elevation requires more time. Boiling point lowered: water doesn't get as hot.
 - If pressure too low, water goes directly from ice to vapor (like dry ice CO₂)
- On the other hand, high pressure may make life more difficult to form.
- In addition, the range of temperature for Earth based complex life is less than 325K.



Water: Our Liquid Friend



- A very good temperature buffer
 - Absorbs significant heat before its temperature changes
 - When it vaporizes, it takes heat with it, cooling its original location
- It floats.
 - Good property for life in water.
 - Otherwise, a lake would freeze bottom up, killing life.
 - By floating to the surface, it can insulate the water somewhat.



Keeping It Warm, but not too Warm 🧊

- What controls a planet's temperature?
 - The amount of light received from its star.
 - The amount of energy the planet reflects back.
 - And any Greenhouse effects of the planet.



Greenhouse Explained



Keeping It Warm, but not too Warm

- Earth's greenhouse effect raises the temperature by about 15%.
- Given a star's luminosity, a range of acceptable temperatures translates into a range of distances to the star.
- This range is called the star's habitable zone (HZ), as



• Only a rough guideline.

http://www.solcomhouse.com/Greenhouse_Effect.gif

Habitable Zones– Are you in the Zone?

- Long living star
- Planets with stable orbits (thus stable temps)
- Liquid Water
- Heavy Elements- C, N, O, etc.
- Protection from UV radiation





Galactic Habitable Zone



- Likewise the galaxy has regions that are better suited to life.
- In the inner regions of our galaxy, supernovae are too frequent.
- In the outer regions, there are too few metals.



http://astronomy.swin.edu.au/GHZ/GHZmovie.html

Question

The Greenhouse effect

- a) will destroy our planet.
- b) will hopefully stop this crazy winter.
- c) keeps the Earth warmer than it would be otherwise at its distance from the Sun.
- d) is all Man-Made.
- e) keeps the Earth colder than it would be otherwise at its distance from the Sun.

The Sun's Variation

- As the Sun ages, it gets slightly brighter.
- When it was younger, its luminosity was 70% current values.
- A young Earth should have been 20K coldericeball!
 - During our ice ages, the temperature only changed by about 1%!



http://www.cherishclaire.com/iceball.htm

The Sun's Variation

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- There is evidence that the Earth did nearly freeze over- 2.8 billion years ago and 700 million years ago.
- Probably changes in the Greenhouse gases.
- This implies that the habitable zone can vary with time, thus the real habitable zone is smaller than shown before?
- Some have postulated that real zone is only 0.95 to 1.01 AU! If the Earth were 1% farther away– Iceballed. And n_p would be very small ~ 0.1.



http://www.soest.hawaii.edu/gerard/GG108/images/bylot.jpg

Earth's Atmosphere: Trapping CO₂ for Fun and Profit

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- Most recent studies suggest an efficient planet negative-feedback mechanism (like a thermostat).
 - CO₂ cycles from atmosphere (greenhouse gas) and oceans (buried sediment especially carbonate rock).
 - CO₂ in atmosphere: temporarily dissolved CO₂ in rainfall reacts with weathered rocks, trapping it.



Earth's Atmosphere: Trapping CO₂ for Fun and Profit



- Negative feedback process
 - Increase in temperature: evaporation of oceans, more rainfall, more weathering and CO_2 reduction, so decrease in temperature.
 - This negative feedback stabilizes the Earth's temperature.



Life Adds to Feedback

• Life increases the

weathering of rock.

planet temperature.

• Regardless, the negative

around 1- more Earth

chauvinism?



Life in the Solar System?



- We want to examine in more detail the backyard of humans.
- What we find may change our estimates of n_e.





What We Used to Think



Venus must be hotter, as it is closer the Sun, but the cloud cover must reflect back a large amount of the heat.

In 1918, a Swedish chemist and Nobel laureate concluded:

- Everything on Venus is dripping wet.
- Most of the surface is no doubt covered with swamps.
- The constantly uniform climatic conditions result in an entire absence of adaptation to changing exterior conditions.
- Only low forms of life are therefore represented, mostly no doubt, belonging to the vegetable kingdom; and the organisms are nearly of the same kind all over the planet.

http://www.daviddarling.info/encyclopedia/V/Venuslife.html

Turns Out that Venus is Hell

- The surface is hot enough to melt lead
- There is a runaway greenhouse effect
- There is almost no water
- There is sulfuric acid rain
- Not a place to visit for Spring Break.



Our "Twin"

- Always covered in thick clouds of CO₂, which make it the hottest planet in the Solar System.
- Pressure on surface is 90 times that on Earth–like 1 km under the sea



http://antwrp.gsfc.nasa.gov/apod/ap960923.html

Our "Twin"

- Often called the morning star or the evening star.
 3rd brightest object in the sky.
- Often mistaken for a UFO.
- Retrograde rotation Sun rises in west
- No moons, no magnetic field



http://antwrp.gsfc.nasa.gov/apod/ap960923.html



USSR Academy of Sciences / Brown University

Mostly Basalts-like rocks, indicative of volcanoes

The Venusian Surface Revealed

- We can't see Venus' surface in visible light, clouds block the view
- Magellan's Radar showed the surface
- Most of surface is smooth lava flows
- Many large volcanoes
- Probable ongoing volcanism



Surface of Venus: Radar



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Impacts on Venus

- Venus has about 1,000 craters, often clustered
- No trace of heavy bombardment
- Cratering rate indicates Venus' surface about 500 million yrs old
- Why?
 - Possibility: Extreme temperatures soften rock, making the surface subject to catastrophic volcanic upheaval

Venus' Interior

- Venus' size and density are roughly equal to Earth's
 - Indicates iron core of similar size
- No magnetic field
 Very slow rotation -243 Earth days



Runaway Greenhouse

- On Earth, greenhouse gasses insulate us
 - Keep Earth 35 K warmer than it would be otherwise
- On Venus, massive amounts of CO₂ keep it incredibly hot
 - Almost 300 K warmer!
 - The hottest planet in the Solar System



What Happened to Venus?

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- It really should have been more like Earth, but the atmosphere is much different.
- Earth's atmosphere is mostly O₂ from life, but early Earth was N.
- Earth and Venus have similar amounts of carbon & nitrogen, but...



http://www.digitalart.ab.ca/art/ren/images/birth-of-venus.jpg

Why So Different?

- Earth's carbon is locked up
 - Dissolved in the oceans
 - Locked into rocks and life



- Venus' carbon is in its atmosphere
 - Too close to the Sun for liquid water
 - No oceans to trap the carbon dioxide
 - No life to process the carbon into sedimentary rocks

http://www.edgechaos.com/MECA/WALLART/VR89/venus.jpeg

What Happened to Venus?

- Apparently Venus lost its H₂O- no oceans and no sediments.
- Probably the atmospheric temperature was hot enough for water to travel high enough to be broken apart by UV radiation, the H was lost and the O reacted with something else.
- Irreversible procedure!
- Which is why greenhouse effect is worrisome here too!
- The Earth traps water vapor in the cool tropopause at 14km.



http://photos1.blogger.com/blogger/4103/1148/1600/Venus%20Wimbeldon05.jpg