

Astronomy 330



This class (Lecture 11):

Life on Earth

Daniel Miller-McLemore
Ali Timm

Next Class:

Life in the Solar System

Daniel Borup

Music: *Jesus Came From Outta Space*– Supergrass

HW 2



- **Nicholas Cox**
<http://www.ufoevidence.org/>
- **Rebecca Reizner**
<http://www.alien-ufo-pictures.com/>

Presentations



- **Daniel Miller-McLemore**
[Faster than Light Travel](#)
- **Ali Timm**
[Area 51](#)

Outline



- n_e
- Why is Earth a good place for life?
- What about Venus or Mars?

Drake Equation

Frank Drake



That's 7.5 planetary systems/year



$$N = R_* \times f_p \times n_e \times f_l \times f_i \times f_c \times L$$

# of advanced civilizations we can contact in our Galaxy today	Star formation rate	Fraction of stars with planets	# of Earthlike planets per system	Fraction on which life arises	Fraction that evolve intelligence	Fraction that communicate	Lifetime of advanced civilizations
10 stars/yr	10 stars/yr	0.75 systems/star	planets/system	life/planet	intel./life	comm./intel.	yrs/comm.

n_e



Complex term, so let's break it into two terms:

- n_p : number of planets suitable for life per planetary system
- f_s : fraction of stars whose properties are suitable for life to develop on one of its planets

<http://mike.cecs.csulb.edu/~kjlivio/Wallpapers/Planets%2001.jpg>

$$n_e = n_p \times f_s$$



Moon Impact on Life?



- Some think that our large Moon is very important for life on Earth.
 - Tides! Important to move water in and out of pools.
 - Stable Axial Tilt: 23.5 deg offset from the collision
 - Metals! Heavy elements at Earth's surface may be from core of impactor.

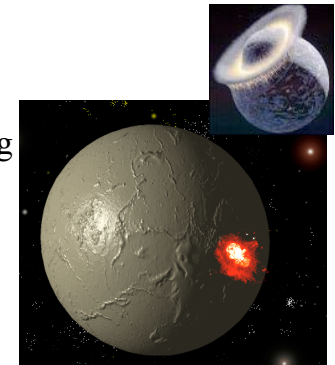


http://www.michaelbach.de/ot/size_moon/index.html

Implications



- **Hot, Hot, Hot!** Even if the moon theory is incorrect, other smaller bodies were playing havoc on the surface.
- When they impact, they release kinetic energy and gravitational potential.
- In addition, some of the decaying radioactive elements heated up the Earth— stored supernova energy!
- The planetesimals melt, and the Earth went through a period of differentiation.



<http://www.udel.edu/Biology/Wags/wagat/worldspage/impact.gif>

Early Earth



- No atmosphere
- No water
- High temp
- No life.....
- Big rocks keep falling on my head...



<http://www.black-cat-studios.com/catalog/earth.html>

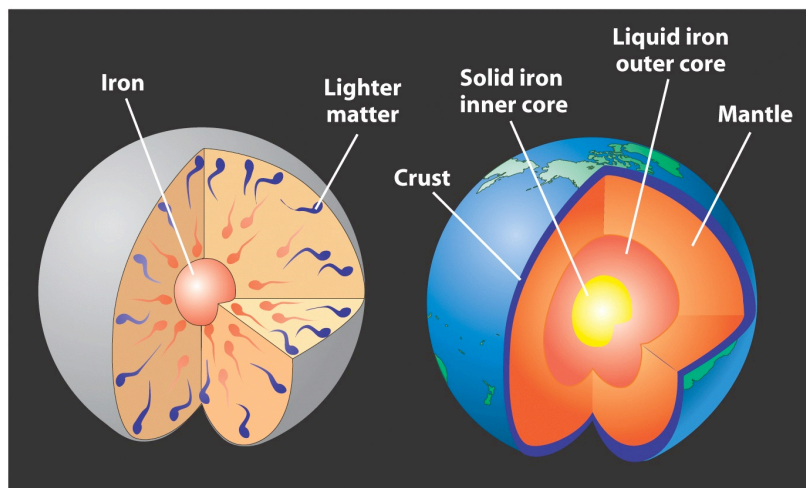
Question



Which of the following does NOT well describe the early Earth?

- So hot that the surface had molten rock.
- There was no water.
- The surface kept getting hit by really, really big rocks.
- The oxygen rich atmosphere caused quick oxidation (rusting) of iron-rich rocks
- No chance of life at this stage.

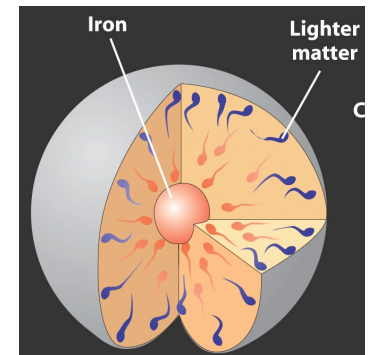
Planetary Differentiation



Differentiation: Iron Catastrophe



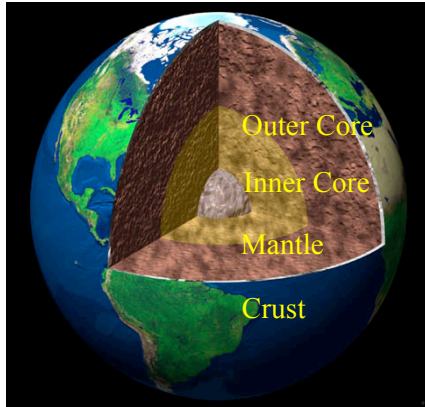
- Average density of Earth is 5.5 g/cm^3
- Average density on the surface is 3 g/cm^3
- So, something heavy must be inside
- When the Earth formed it was molten
 - Heavy materials (e.g. iron, nickel, gold) sank
 - Lighter materials (e.g. silicon, oxygen) floated to the top



Structure



- Luckily, not all of the iron sank to the center, else we would be still in the Stone Age.
- Temperature increases as you go deeper underground. From around 290 K on surface to nearly 5000 K at center.
 - Heated by radioactive decay
 - Supernovae remnants
- Earth's magnetic field is established early on.. after the iron catastrophe... good for life.



The Crust



- Outside layer of the Earth (includes oceans) floats on top of still hot interior
 - About 50 km thick
 - Coldest layer – rocks are rigid
- Mostly silicate rocks
 - Made of lighter elements like silicon, oxygen, and aluminum
- Oxygen and water are abundant
- Excellent insulator
 - Keeps the Earth's geothermal heat inside!



Today's Earth Surface



- 70% of the Earth's surface is covered with water
 - Ocean basins
 - Sea floors are young, none more than 200 million years old
- 30% is dry land – Continents
 - Mixture of young rocks and old rocks
 - Up to 4.2 billion years old



Geologically Active Surface



- The young rocks on the Earth's surface indicate it is geologically active
- Where do these rocks come from?
 - Volcanoes
 - Rift valleys
 - Oceanic ridges
- Air, water erode rocks
- **The surface is constantly changing**



Recycling Bio-elements



- From gravity and radioactivity, the core stays hot.
- This allows a persisting circulation of bioelements through continental drift— melting of the crust and re-release through volcanoes.
- Otherwise, certain elements might get locked into sediment layers— e.g. early sea life.
- Maybe planets being formed now, with less supernovae, would not have enough radioactivity to support continental drifts and volcanoes. (Idea of Peter Ward and Donald Brownlee.)

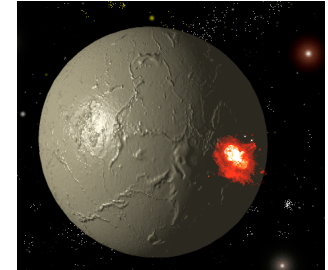


<http://www.pahala-hawaii.com/j-page/image/activevolcanoe.jpg>

The Earth's 1st Atmosphere



- The inner disk had most gases blown away and the proto-Earth was not massive enough to capture these gases.
- Any impacts (e.g. the moon), would have blown any residual atmosphere away.
- The first atmosphere was probably H and He, which was lost quickly.



<http://www.udel.edu/Biology/Wags/wagant/worldspage/impact.gif>

The Earth's 1st Atmosphere



- The interior heat of the Earth helped with the Earth's early atmosphere.
- Volcanoes released gases (water vapor and CO₂)
- Another scenario is that impacted comets released – water (H₂O), carbon dioxide (CO₂), and Nitrogen (N₂) – the first true atmosphere.
- The water condensed to form the oceans and much of the CO₂ was dissolved in the oceans and incorporated into sediments— such as calcium carbonate (CaCO₃).



<http://www.fli-cam.com/images/comet-liner.jpg>

Our Atmosphere



- Rocks with ages greater than 2 billion years show that there was little or no oxygen in the Earth's atmosphere.
- The current composition: 78% nitrogen, 21% oxygen, and trace amounts of water, carbon dioxide, etc.
- Where did the oxygen come from?
- Cyanobacteria made it.
 - Life on Earth modifies the Earth's atmosphere.

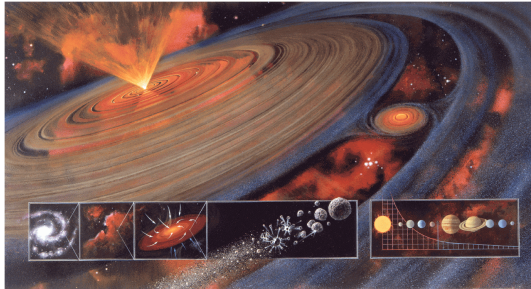


<http://www.uweb.ucsb.edu/~rixfury/conclusion.htm>

This New Planet



- Mostly oceans and some solid land (all volcanic).
- Frequent impacts of remaining planetesimals (ending about 3.8 billion years ago).
- Impacts would have sterilized the young Earth— Mass extinctions and maybe vaporized any oceans (more comets?).

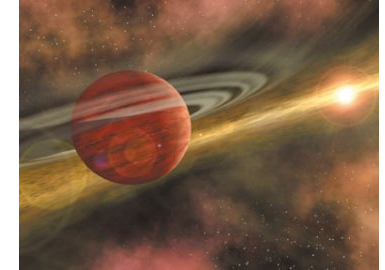


<http://www.agnd.uni-potsdam.de/~frank/Images/painting.gif>

This New Planet



- Impacts and volcanic activity created the continental landmasses.
- Little oxygen means no ozone layer— flooded with ultraviolet light on surface.
- Along with lightning, radioactivity, and geothermal heat, provided energy for chemical reactions.
- **BUT, life on the surface not possible!**



Question



The Earth's first atmosphere was

- a) much like today's atmosphere, but older.
- b) Trick Question. There was no atmosphere.
- c) likely just H and He, and blown away quickly.
- d) made from comets.
- e) a combination of volcano gases and comet collisions.



Water

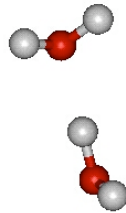


- Water is a key to life on Earth.
- Primary constituent of life— “Ugly bags of mostly water”
 - Life is about 90% water by mass.
- Primary role as a solvent
 - Dissolves molecules to bring nutrients and remove wastes. Allows molecules to “move” freely in solution.
 - Must be in liquid form, requiring adequate pressure and certain range of temperatures.
- This sets a requirement on planets, if we assume that all life requires water.

Water as a Solvent



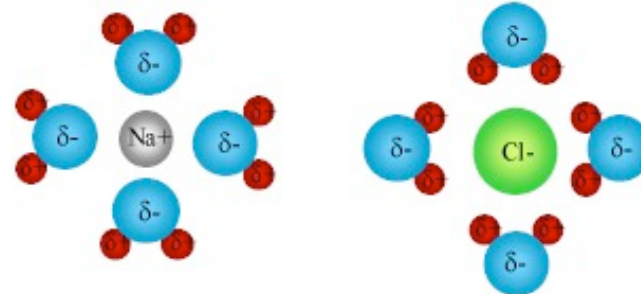
- The water molecule is “polar”. The oxygen atoms have more build-up of negative charge than the hydrogen. This allows water molecules to link up, attracted to each other.
- In this way, water attracts other molecules, surrounds them and effectively dissolves them into solution.



Example: Dissolving Table Salt



The partial charges of the water molecule are attracted to the Na^+ and Cl^- ions. The water molecules work their way into the crystal structure and between the individual ions, surrounding them and slowly dissolving the salt.

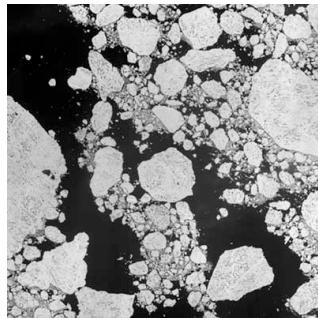


http://www.visionlearning.com/library/module_viewer.php?mid=57

Water: Our Liquid Friend



- A very good temperature buffer
 - Absorbs significant heat before its temperature changes
 - When it vaporizes, it takes heat with it, cooling its original location
- It floats.
 - Good property for life in water.
 - Otherwise, a lake would freeze bottom up, killing life.
 - By floating to the surface, it can insulate the water somewhat.



Keeping it Useful: Atmosphere



- Need to have enough pressure to keep water from boiling away at low temperature
 - Cooking at higher elevation requires more time. Boiling point lowered: water doesn't get as hot.
 - If pressure too low, water goes directly from ice to vapor (like dry ice CO_2)
- On the other hand, high pressure may make life more difficult to form.
- In addition, the range of temperature for Earth based complex life is less than 325K.

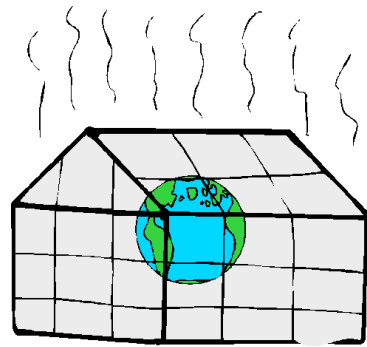


http://www.astro.su.se/~magnus/large/Boiling_water.jpg

Keeping It Warm, but not too Warm

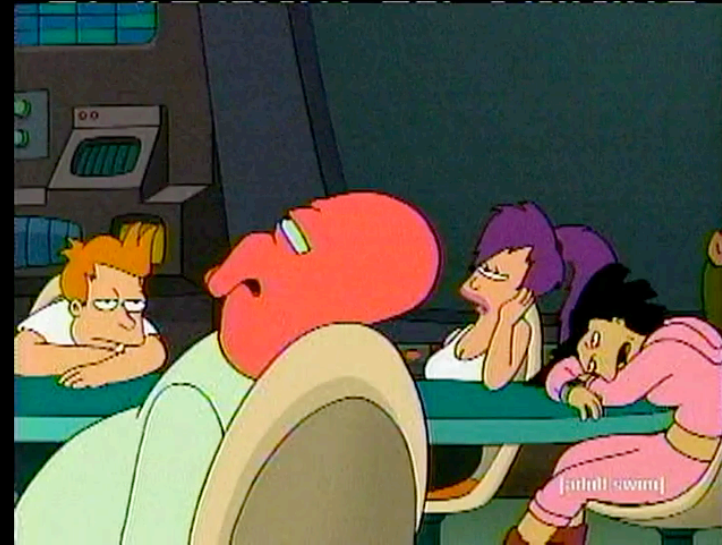


- What controls a planet's temperature?
 - The amount of light received from its star.
 - The amount of energy the planet reflects back.
 - And any Greenhouse effects of the planet.



http://www.solcomhouse.com/Greenhouse_Effect.gif

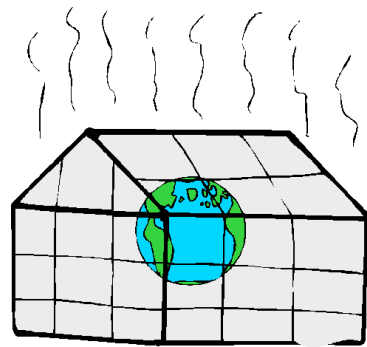
Greenhouse Explained



Keeping It Warm, but not too Warm



- Earth's greenhouse effect raises the temperature by about 15%.
- Given a star's luminosity, a range of acceptable temperatures translates into a range of distances to the star.
- This range is called the star's habitable zone (HZ), as planets in this range have temperatures suited for life.
- Only a rough guideline.

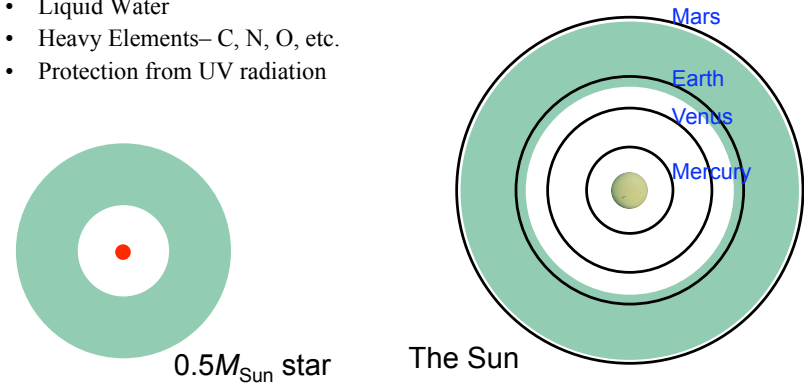


http://www.solcomhouse.com/Greenhouse_Effect.gif

Habitable Zones— Are you in the Zone?



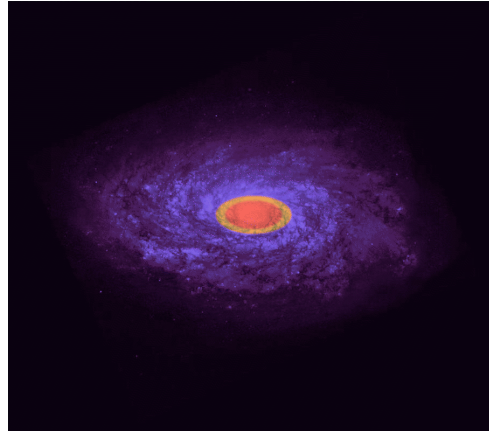
- Long living star
- Planets with stable orbits (thus stable temps)
- Liquid Water
- Heavy Elements—C, N, O, etc.
- Protection from UV radiation



Galactic Habitable Zone



- Likewise the galaxy has regions that are better suited to life.
- In the inner regions of our galaxy, supernovae are too frequent.
- In the outer regions, there are too few metals.



<http://astronomy.swin.edu.au/GHZ/GHZmovie.html>

Question



The Greenhouse effect

- a) will destroy our planet.
- b) will hopefully stop this crazy winter.
- c) keeps the Earth warmer than it would be otherwise at its distance from the Sun.
- d) is all Man-Made.
- e) keeps the Earth colder than it would be otherwise at its distance from the Sun.

The Sun's Variation



- As the Sun ages, it gets slightly brighter.
- When it was younger, its luminosity was 70% current values.
- A young Earth should have been 20K colder– iceball!
- During our ice ages, the temperature only changed by about 1%.



<http://www.cherishclaire.com/iceball.htm>

The Sun's Variation



- There is evidence that the Earth did nearly freeze over– 2.8 billion years ago and 700 million years ago.
- Probably changes in the Greenhouse gases.
- This implies that the habitable zone can vary with time, thus the real habitable zone is smaller than shown before?
- Some have postulated that real zone is only 0.95 to 1.01 AU! If the Earth were 1% farther away– Iceballed. And n_p would be very small ~ 0.1 .



<http://www.soest.hawaii.edu/gerard/GG108/images/bylot.jpg>

Earth's Atmosphere: Trapping CO₂ for Fun and Profit



- Most recent studies suggest an efficient planet negative-feedback mechanism (like a thermostat).
 - CO₂ cycles from atmosphere (greenhouse gas) and oceans (buried sediment especially carbonate rock).
 - CO₂ in atmosphere: temporarily dissolved CO₂ in rainfall reacts with weathered rocks, trapping it.
- Negative feedback process
 - Increase in temperature: evaporation of oceans, more rainfall, more weathering and CO₂ reduction, so decrease in temperature.
 - This negative feedback stabilizes the Earth's temperature.



<http://www.wildtech.org/images/feedback.gif>

Life Adds to Feedback



- Life increases the weathering of rock.
- J.E. Lovelock has proposed that life also stabilizes the planet temperature.
- Regardless, the negative feedback helps with the habitable zone, so we can estimate perhaps n_p is more around 1– more Earth chauvinism?



While testing out his new cereal mix on his horse, Dave gets some unexpected feed-back.

<http://www.cts.com/~borderln/>