

# ASTR 150

- ▶ **Homework 1** due Monday
- ▶ **Planetarium** shows begin next week
  - ▶ Need to **register** to reserve a spot
  - ▶ Registration, schedule, and report info on course website
- ▶ **Register** your iClicker!
- ▶ Last time: Meteors
- ▶ Today: Asteroids



"Be realistic! - how can lumps of rock zooming around in space affect our lives down here?"

**Music: *Asteroid* – Killing Joke**

# i>clicker question

The current Compass does not allow easy access to HW grades for each attempt. I propose a new solution. I can set it up so that Compass tells you which questions you missed, but then you only get 3 chances!

- A.** Yes, let's change it. This will make it more easy to learn what I missed and allow me to learn more efficiently.
- B.** No, let's not change it. I like having unlimited attempts.
- C.** I don't care either way.

# The World of Atoms



“If, in some cataclysm, all of scientific knowledge were to be destroyed, and only one sentence passed on to the next generation of creatures, what statement would contain the most information in the fewest words? I believe it is the **atomic hypothesis** (or the **atomic fact**, or whatever you want to call it) that **all things are made of atoms**--little particles that move around in perpetual motion, attracting each other when they are a little distance apart, but repelling upon being squeezed into one another. In that one sentence, you will see, there is an enormous amount of information about the world, if just a little imagination and thinking are applied.”

--Richard Feynman

<http://www.youtube.com/watch?v=v3pYRn5j7oI>

[http://www.nobelprize.org/nobel\\_prizes/physics/laureates/1965/feynman.html](http://www.nobelprize.org/nobel_prizes/physics/laureates/1965/feynman.html)

# Jiggling

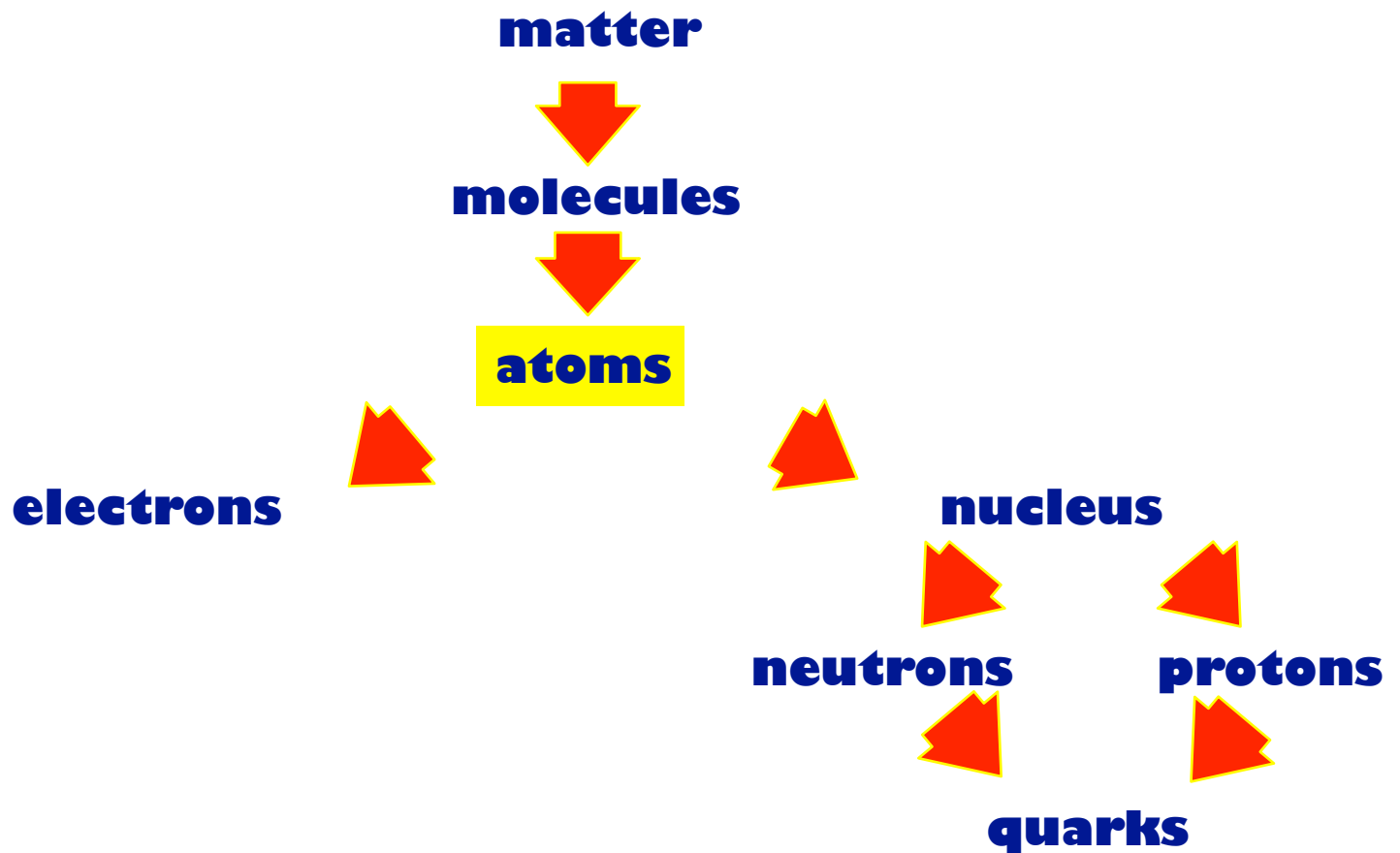


--Richard Feynman

<http://www.youtube.com/watch?v=v3pYRn5j7oI>

# Matter\*

All known substances ever studied in any lab have this structure



*\*Weirdo dark matter not included in this discussion*

# iClicker Poll:

## Atoms in a Grain of Sand

Think of the smallest grain of sand  
between your toes at the beach



Go with your gut, vote your conscience!  
Pick the largest answer that works

**The number of atoms in 1 grain of sand is:**

- A. more than the number of people in this room
- B. more than capacity of sold-out Memorial Stadium
- C. more than the Chicagoland population
- D. more than the population of Illinois
- E. more than population of planet Earth



**Answer:**

- ▶ **1 grain of sand** is made of about  **$10^{19}$  atoms**
- ▶ That is: **10,000,000,000,000,000,000 atoms**

**compare: global population  $\approx 6.5 \times 10^9$  people**

**lesson: atoms are numerous and tiny!**

# The Structure of Atoms

At the atom's center: a single **nucleus**

In orbit around nucleus: one or more **electrons**

electron:

- ▶ electric charge -1

nucleus:

- ▶ made of protons: charge +1
- ▶ and neutrons: charge 0

total charge of atom: set by # electrons

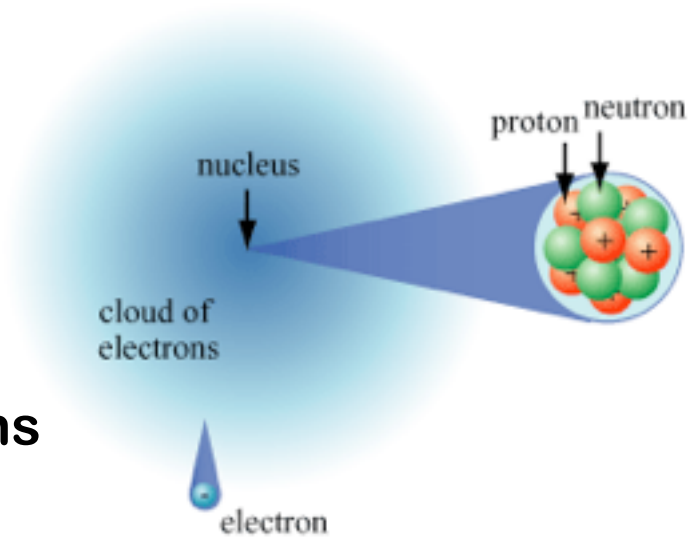
- ▶ if  $\# e = \# p$  → atom is **neutral**
- ▶ if  $\# e = \# p - 1$  → charge = +1: atom is (singly) **ionized**

masses:  $m_p \approx m_n \approx 2000m_e$

- ▶ most of atom **mass** is in dense **nucleus**
- ▶ most of atom **volume** occupied by **electron orbits**

**e** moves around nucleus

*Q: what does this tell us about forces in atoms?*





# Forces in Atoms: Chemistry

## Electron orbits: curved paths

- ▶ Motion must be accelerated
- ▶ → needs to be a net force
- ▶ And there is! nucleus & electrons attracted by **electric force**

Rule: **opposite charges attract, like charges repel**

- ▶ Atom structure similar to Solar System: attractive force → orbits
- ▶ Big object in center, orbiting smaller objects

Charge of nucleus = # p

- ▶ Sets **force** on e → e **orbit** properties
- ▶ Determines how atoms interact: **chemistry!**
- ▶ 92 atom varieties = **elements**
- ▶ From hydrogen = 1p to uranium = 92p

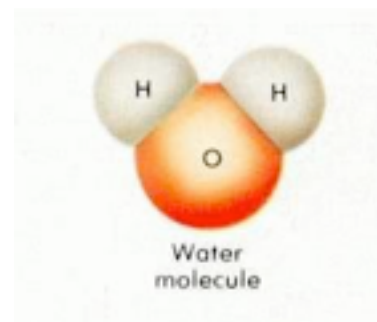
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|----|----|----|----|----|----|----|----|----|----|----|----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|----|----|-----|-----|-----|-----|
| 1  |    |    |    |    |    |    |    |    |    |    |    |     |     |     |     |     | 2   |     |     |     |     |     |     |     |     |     |    |    |     |     |     |     |
| H  |    |    |    |    |    |    |    |    |    |    |    |     |     |     |     |     | He  |     |     |     |     |     |     |     |     |     |    |    |     |     |     |     |
| 3  | 4  |    |    |    |    |    |    |    |    |    |    | 5   | 6   | 7   | 8   | 9   | 10  |     |     |     |     |     |     |     |     |     |    |    |     |     |     |     |
| Li | Be |    |    |    |    |    |    |    |    |    |    | B   | C   | N   | O   | F   | Ne  |     |     |     |     |     |     |     |     |     |    |    |     |     |     |     |
| 11 | 12 |    |    |    |    |    |    |    |    |    |    | 13  | 14  | 15  | 16  | 17  | 18  |     |     |     |     |     |     |     |     |     |    |    |     |     |     |     |
| Na | Mg |    |    |    |    |    |    |    |    |    |    | Al  | Si  | P   | S   | Cl  | Ar  |     |     |     |     |     |     |     |     |     |    |    |     |     |     |     |
| 19 | 20 | 21 | 22 | 23 | 24 | 25 | 26 | 27 | 28 | 29 | 30 | 31  | 32  | 33  | 34  | 35  | 36  |     |     |     |     |     |     |     |     |     |    |    |     |     |     |     |
| K  | Ca | Sc | Ti | V  | Cr | Mn | Fe | Co | Ni | Cu | Zn | Ga  | Ge  | As  | Se  | Br  | Kr  |     |     |     |     |     |     |     |     |     |    |    |     |     |     |     |
| 37 | 38 | 39 | 40 | 41 | 42 | 43 | 44 | 45 | 46 | 47 | 48 | 49  | 50  | 51  | 52  | 53  | 54  |     |     |     |     |     |     |     |     |     |    |    |     |     |     |     |
| Rb | Sr | Y  | Zr | Nb | Mo | Tc | Ru | Rh | Pd | Ag | Cd | In  | Sn  | Sb  | Te  | I   | Xe  |     |     |     |     |     |     |     |     |     |    |    |     |     |     |     |
| 55 | 56 |    |    |    |    |    |    |    |    |    |    | 72  | 73  | 74  | 75  | 76  | 77  | 78  | 79  | 80  | 81  | 82  | 83  | 84  | 85  | 86  |    |    |     |     |     |     |
| Cs | Ba |    |    |    |    |    |    |    |    |    |    | Hf  | Ta  | W   | Re  | Os  | Ir  | Pt  | Au  | Hg  | Tl  | Pb  | Bi  | Po  | At  | Rn  |    |    |     |     |     |     |
| 87 | 88 |    |    |    |    |    |    |    |    |    |    | 104 | 105 | 106 | 107 | 108 | 109 | 110 | 111 | 112 | 113 | 114 | 115 | 116 | 117 | 118 |    |    |     |     |     |     |
| Fr | Ra |    |    |    |    |    |    |    |    |    |    | Rf  | Db  | Sg  | Bh  | Hs  | Mt  | Ds  | Rg  | Cn  | Uut | Uuq | Uup | Uuh | Uus | Uuo |    |    |     |     |     |     |
|    |    |    |    |    |    |    |    |    |    |    |    |     |     |     |     |     |     | 57  | 58  | 59  | 60  | 61  | 62  | 63  | 64  | 65  | 66 | 67 | 68  | 69  | 70  | 71  |
|    |    |    |    |    |    |    |    |    |    |    |    |     |     |     |     |     |     | La  | Ce  | Pr  | Nd  | Pm  | Sm  | Eu  | Gd  | Tb  | Dy | Ho | Er  | Tm  | Yb  | Lu  |
|    |    |    |    |    |    |    |    |    |    |    |    |     |     |     |     |     |     | 89  | 90  | 91  | 92  | 93  | 94  | 95  | 96  | 97  | 98 | 99 | 100 | 101 | 102 | 103 |
|    |    |    |    |    |    |    |    |    |    |    |    |     |     |     |     |     |     | Ac  | Th  | Pa  | U   | Np  | Pu  | Am  | Cm  | Bk  | Cf | Es | Fm  | Md  | No  | Lr  |

<http://education.jlab.org/itselemental/>

# Chemical Composition

Different elements combine/  
react differently ⇒ **chemistry**

- ▶ Example: water = H<sub>2</sub>O



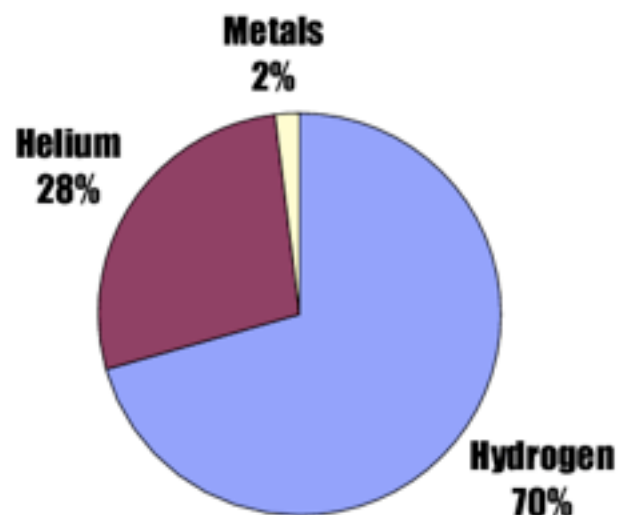
So: “what made of” = “chemical composition”:

- ▶ What kinds of atoms?
- ▶ Which are most, least numerous?

## Examples

- ▶ **Sun, Jupiter:** about 70% hydrogen, 28% helium, 2% other=“metals”
- ▶ **Earth:** about 50% oxygen, 30% silicon, only 0.1% hydrogen

**Solar System Composition by Mass**



# Types of Meteorites: Stonys

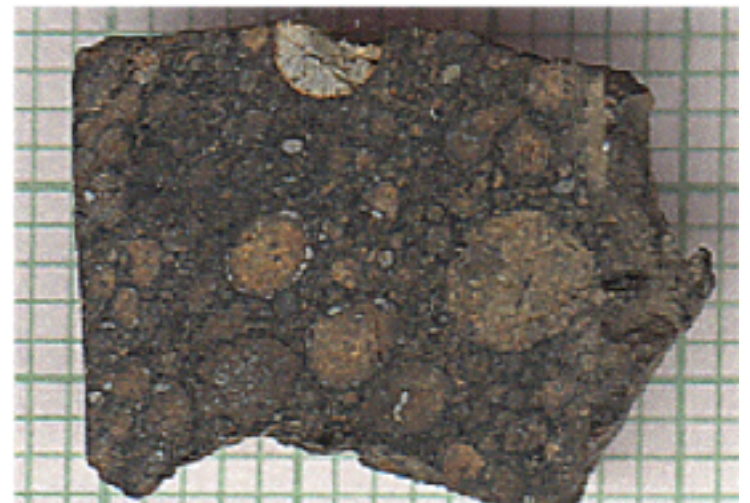
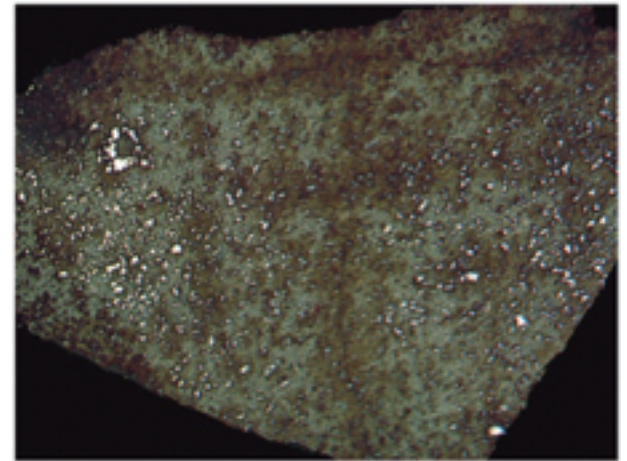
**95-97% of meteorites are stony**

**Made of silicates:  
combinations of silicon and oxygen atoms**

- ▶ Very similar to Earth rocks:  
hard to distinguish

**Many stony meteorites have chondrules**

- ▶ Solidified droplets of ancient material from the early solar system



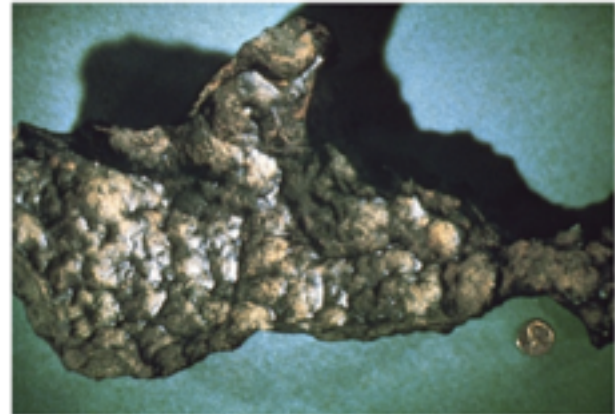
# Types of Meteorites: Irons

2-3% of meteorites are irons  
But, they make up about 40%  
of the meteorites found

*Q: why?*

Easily distinguished from  
Earth rocks

Not pure iron – but iron-nickel  
alloy

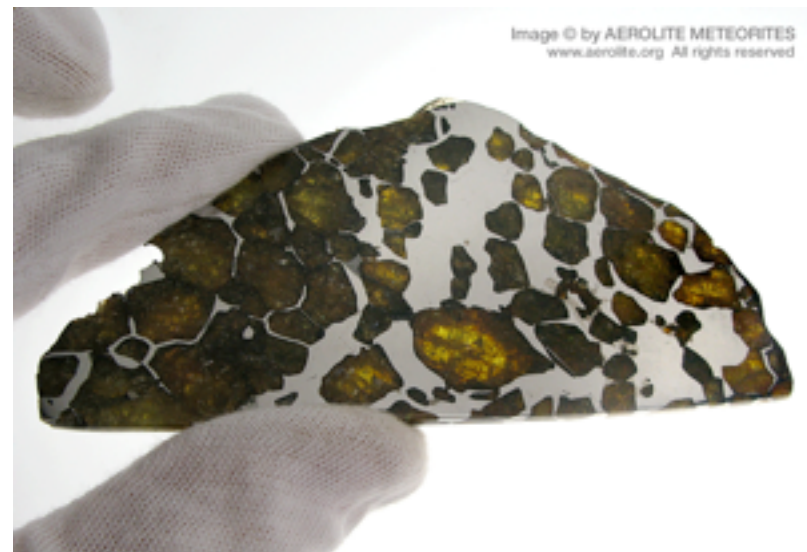
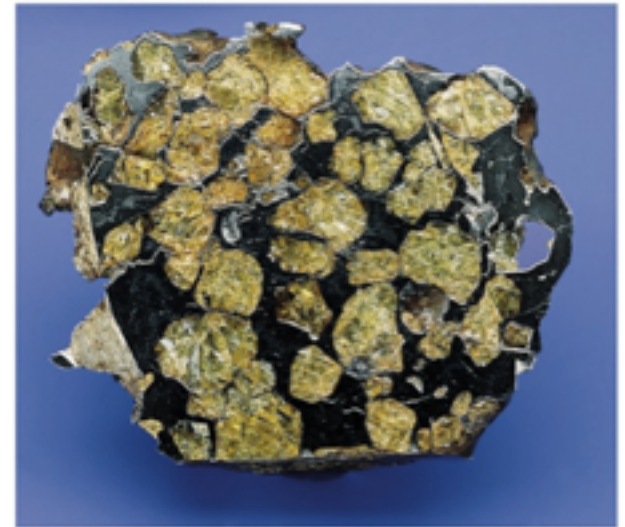


# Types of Meteorites: Stony-Irons

**1% of meteorites are stony-irons**

**Mixture of silicate rock and iron-nickel alloy**

**Often they are fragmental, suggestive of violent processes**



# Largest Meteorite in the World



## The Hoba Meteorite in Namibia

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The largest meteorite ever found is to be found in the grounds of the "Hoba" farm close to Grootfontein. It weighs more than 50 tons, has a volume of 9 cubic metres and hit the ground about 80,000 years ago. The estimated age of meteorite is between 200 and 400 million years. Visitors (vandals) break off bits as souvenirs, so it's gradually shrinking.

# Meteorites from the Moon and Mars

A few meteorites arrive from the **Moon** and **Mars**!

Composition differs from most meteorites

A cheap (but slow) way to acquire moon rocks and Mars rocks



A **Mars** rock found on Earth as a meteorite

*Q: how do we know a meteor came from Mars?*

*Q: how would a piece of Mars get to Earth?*

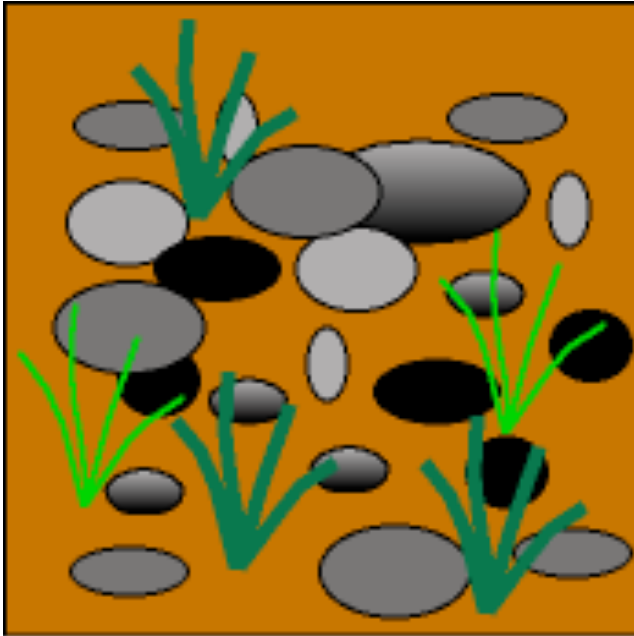
# i>clicker question

You are your friends really want to go looking for meteorites. What would be the best place to look?

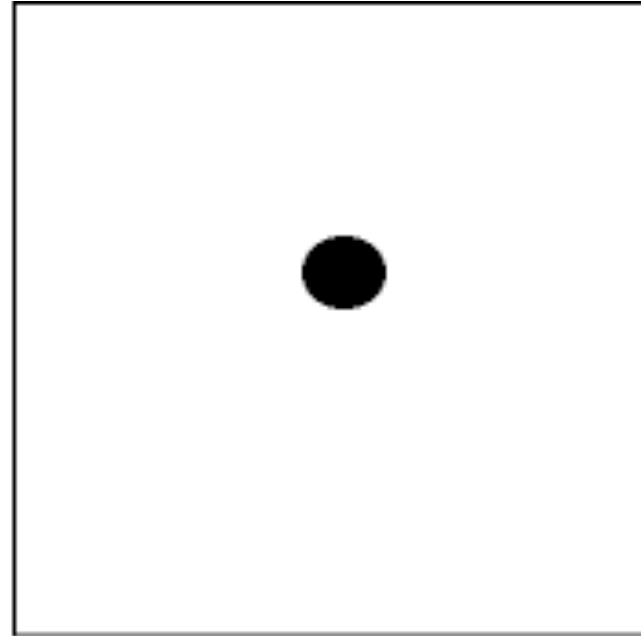
- A. A farm field in Illinois
- B. An urban landscape like Chicago
- C. The Sahara Desert
- D. Antarctica
- E. The ocean floor



# Why Antarctica?



**Typical Earth location**  
Where's the meteorite?



**Antarctica**  
Where's the meteorite?

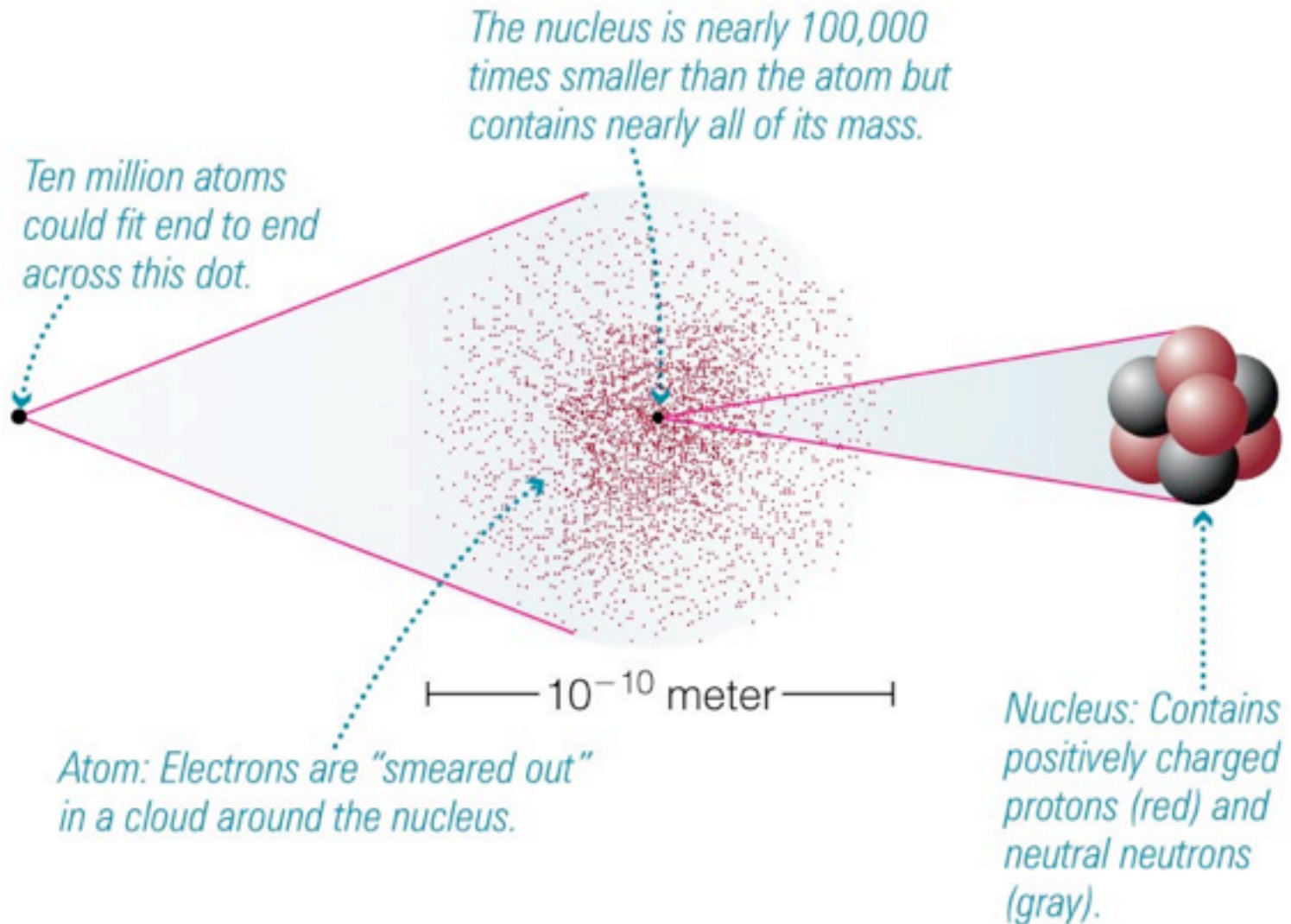
Meteorites stand out against the snow and ice background of Antarctica

# Meteorites are Ancient

- We have found that meteorites are the **oldest objects in the Solar System**
- Oldest meteorites: **age = 4.56 billion years**
- Meteorites tell us the **age of the solar system itself!**

How do we know?

# Inside Atoms: Nuclei



# Radioactive Decay

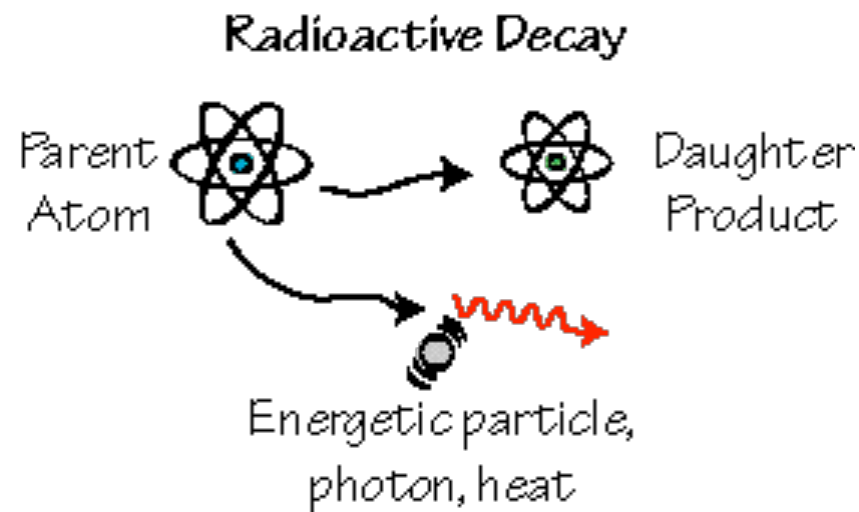
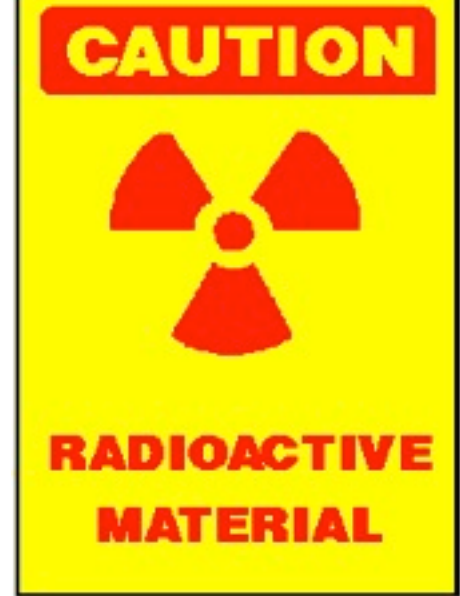
Most atomic nuclei: **stable**

But some nuclei are  
unstable: **radioactive**

After some time,  
**decay**, producing

★ **New nucleus**  
("daughter"):  
different element!  
alchemy!

★ **High-energy particles**  
(electron, sometimes  
photon) that act as  
**heat source**

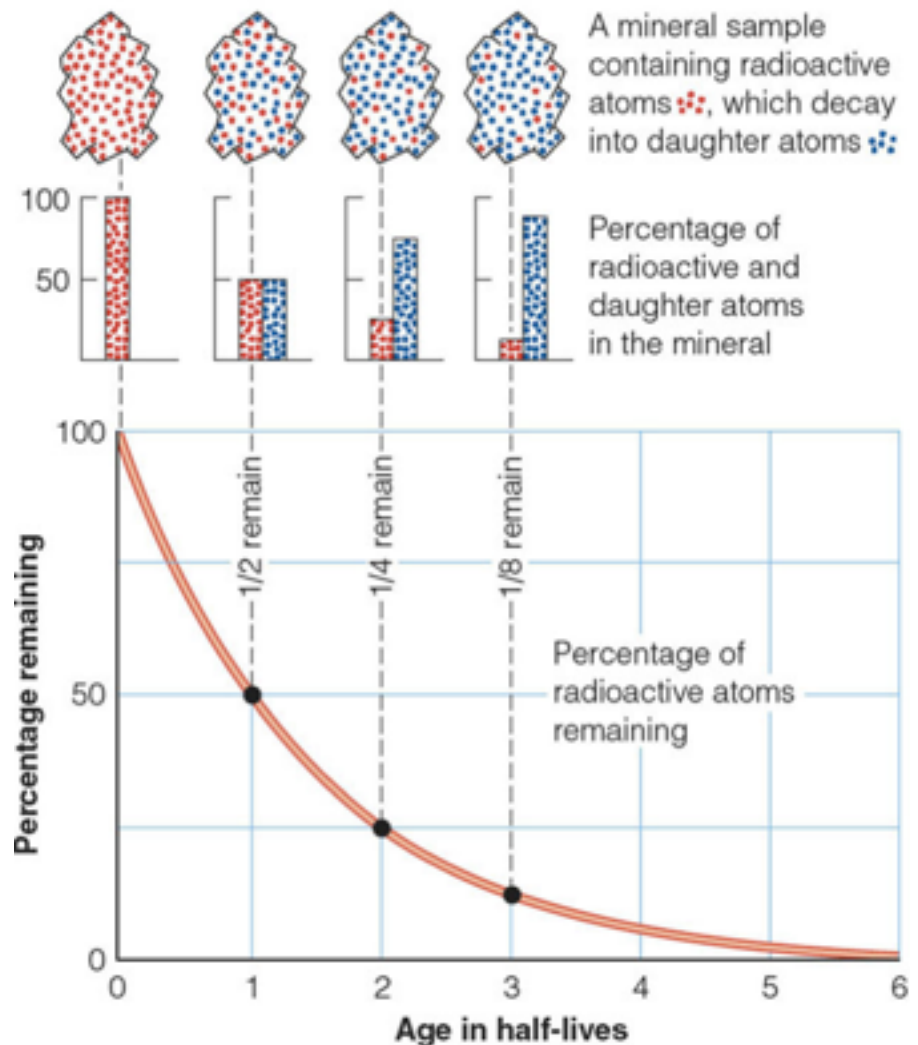


# Radioactivity is a good clock!

As radioactive **parent decays**, the amount of **daughter product increases**

Rate of decay is measured by **half-life**

- ▶ Time it takes for 50% of the radioactive atoms to decay
- ▶ different half-lives for different types of nuclei: some  $\ll$  1 sec, some  $\gg$  age of universe



007 Thomson Higher Education

# Meteorite Dating

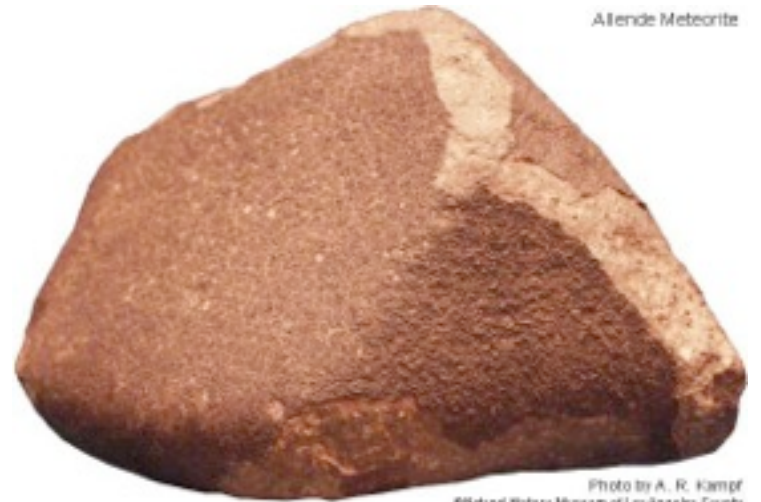
Radioactive “clocks”  
extremely useful!

Procedure:

Collect radioactive nuclei  
from meteor

Measure both parent and  
daughter

Find out how long since  
sample formed!

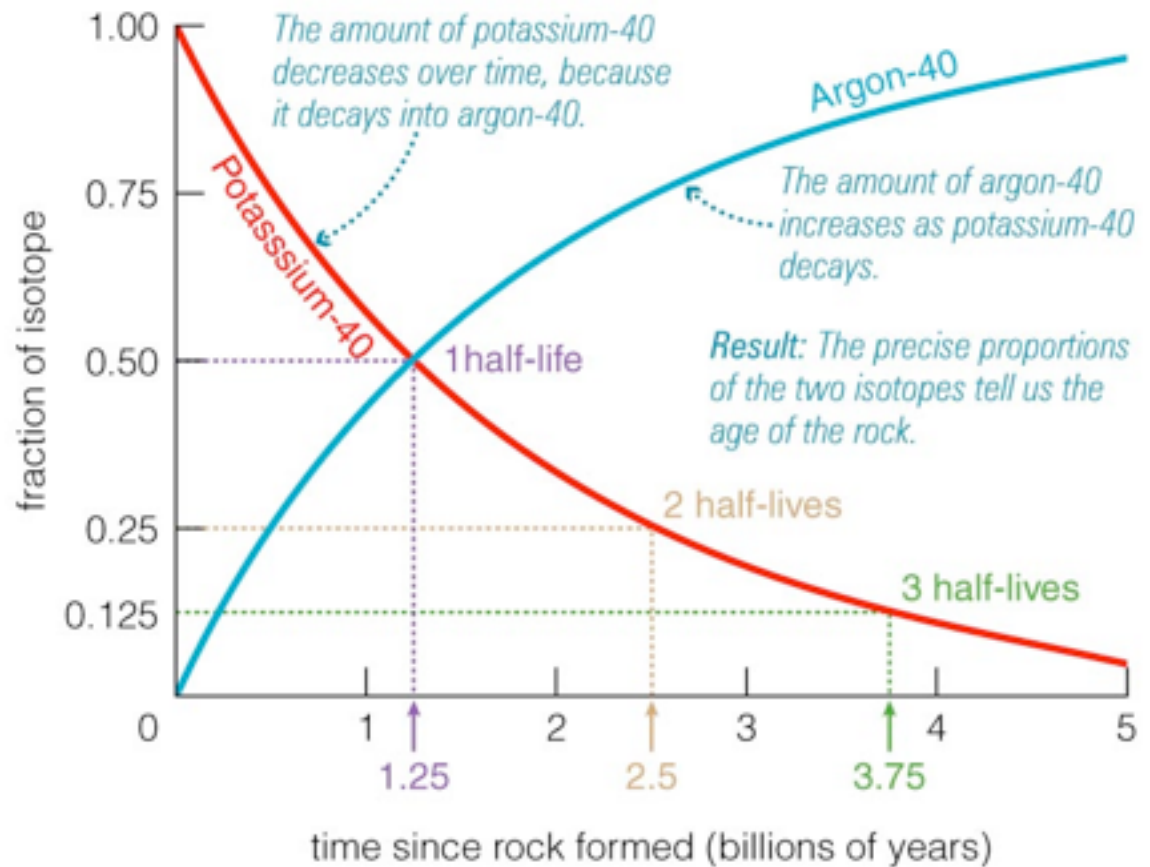


Allende Meteorite

Photo by A. R. Kampf  
©Natural History Museum of Los Angeles County

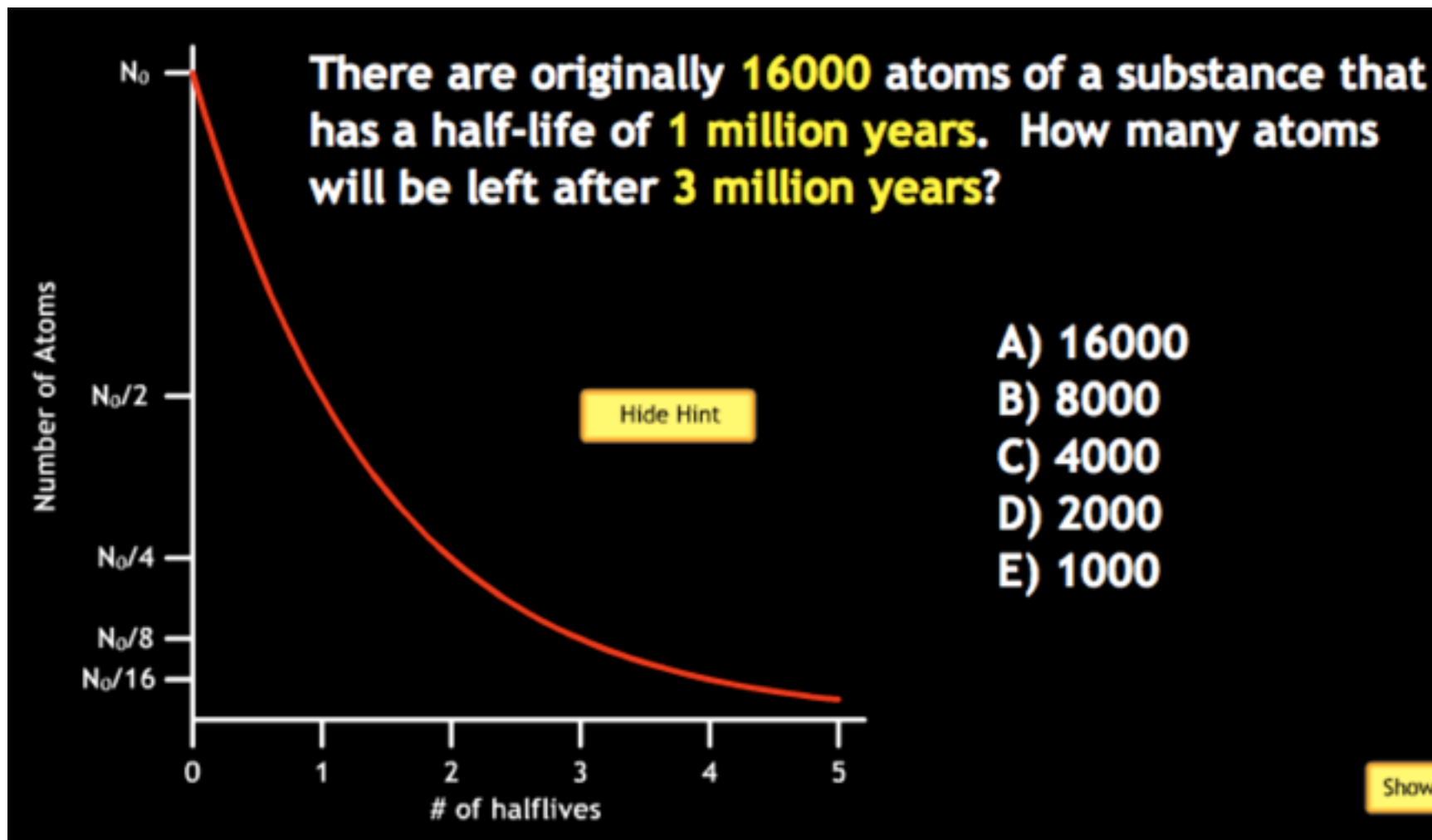
# Example: The Potassium-Argon Method

- Potassium (K)-40 decays to Argon (Ar)-40 with a half-life of 1.25 billion years
- A rock that contains 7 Ar-40 atoms for every 1 K-40 atom is 3 half-lives old or 3.75 billion years old



Potassium-Argon dating has the advantage that the argon is an inert gas that does not react chemically and would not be expected to be included in the solidification of a rock, so any found inside a rock is very likely the result of radioactive decay of potassium. Since the argon will escape if the rock is melted, the dates obtained are to the last molten time for the rock.

# i>clicker question



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3 half-lives  
 $16,000 / 2 = 8000$   
 $8,000 / 2 = 4000$   
 $4000 / 2 = 2000$   
Answer D



# Meteorites are Ancient

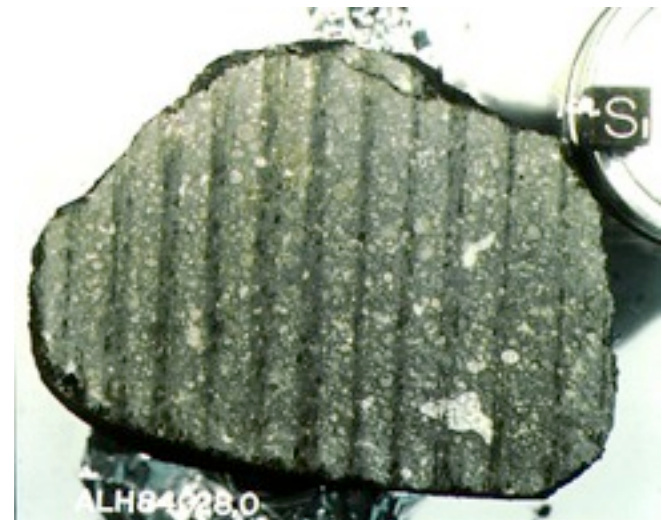
Meteorites are the oldest objects in the Solar System

Oldest are the **carbonaceous chondrites** (a type of stony)

- ▶ Abundant in carbon and water
- ▶ Contain **amino acids** - biochemical ingredients of DNA = **building blocks of life!**
- ▶ 4.56 billion years old

Some have diamonds produced by interstellar shock waves!

Clues to the ancient solar system!



Carbonaceous chondrites

# Who Ordered That?

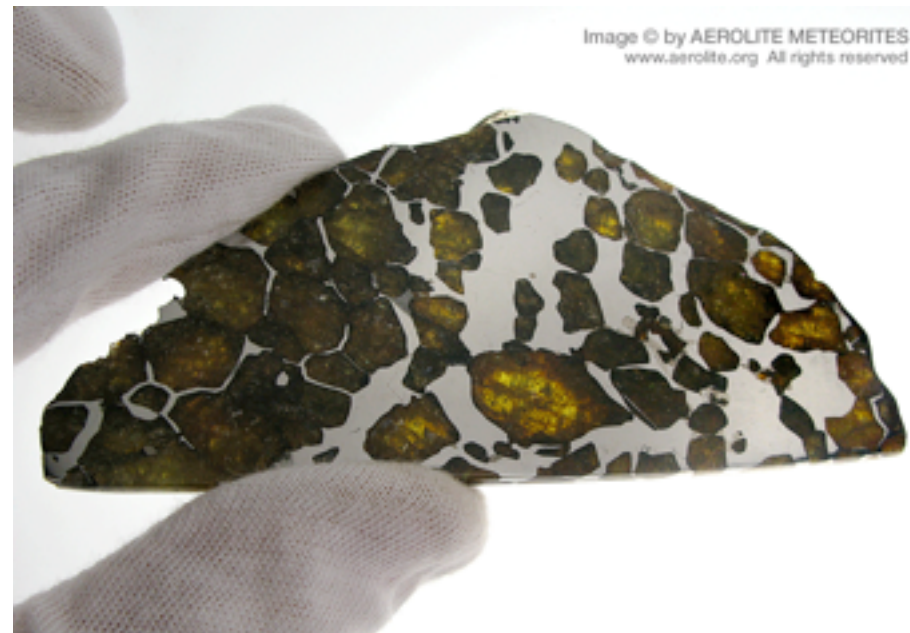
- ▶ **Meteors = extraterrestrial rocks** are falling on our heads!
- ▶ **the sky really is falling!**(occasionally)
- ▶ how did that happen!?

Where do these rocks come from?

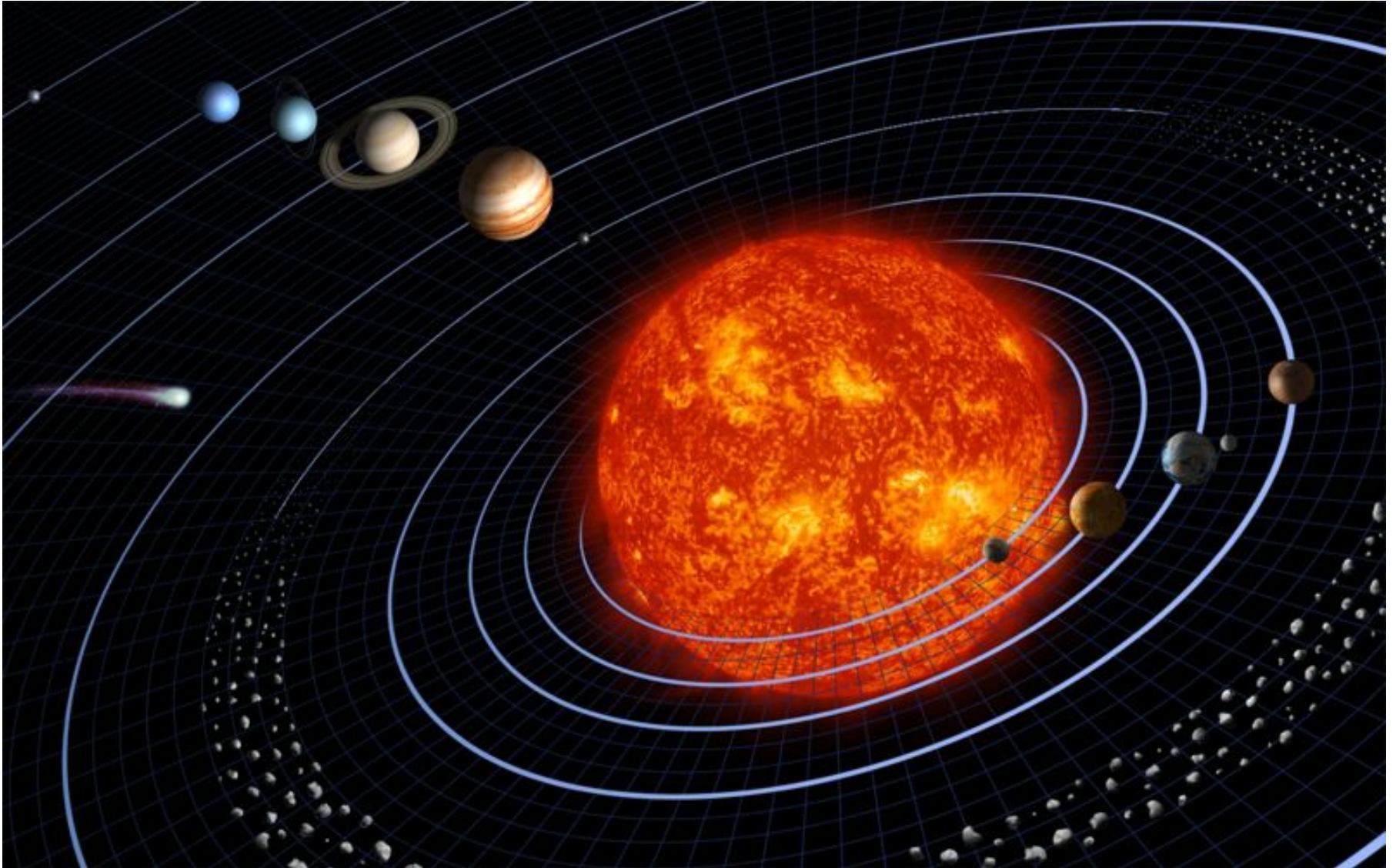
How do they connect to our Solar System?

What do they tell us about the history of our Solar System?

Are we doomed?



# The Solar System



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Ancients knew of the Moon, Sun, Mercury, Venus, Mars, Jupiter, Saturn, comets

“Planet” comes from ancient Greek for wanderer

Comets were once thought to be in the Earth’s atmosphere

Today we know Solar system has:

- 1 star

- 8 planets

- 3+ dwarf planets

- 139+ moons

- countless small solar system bodies

There are currently 11 potential dwarf planets awaiting final classification

\*Small Solar System Bodies: generic name for small bodies like asteroids & comets that orbit the Sun

# Two Types of Solar System Debris



## Asteroids

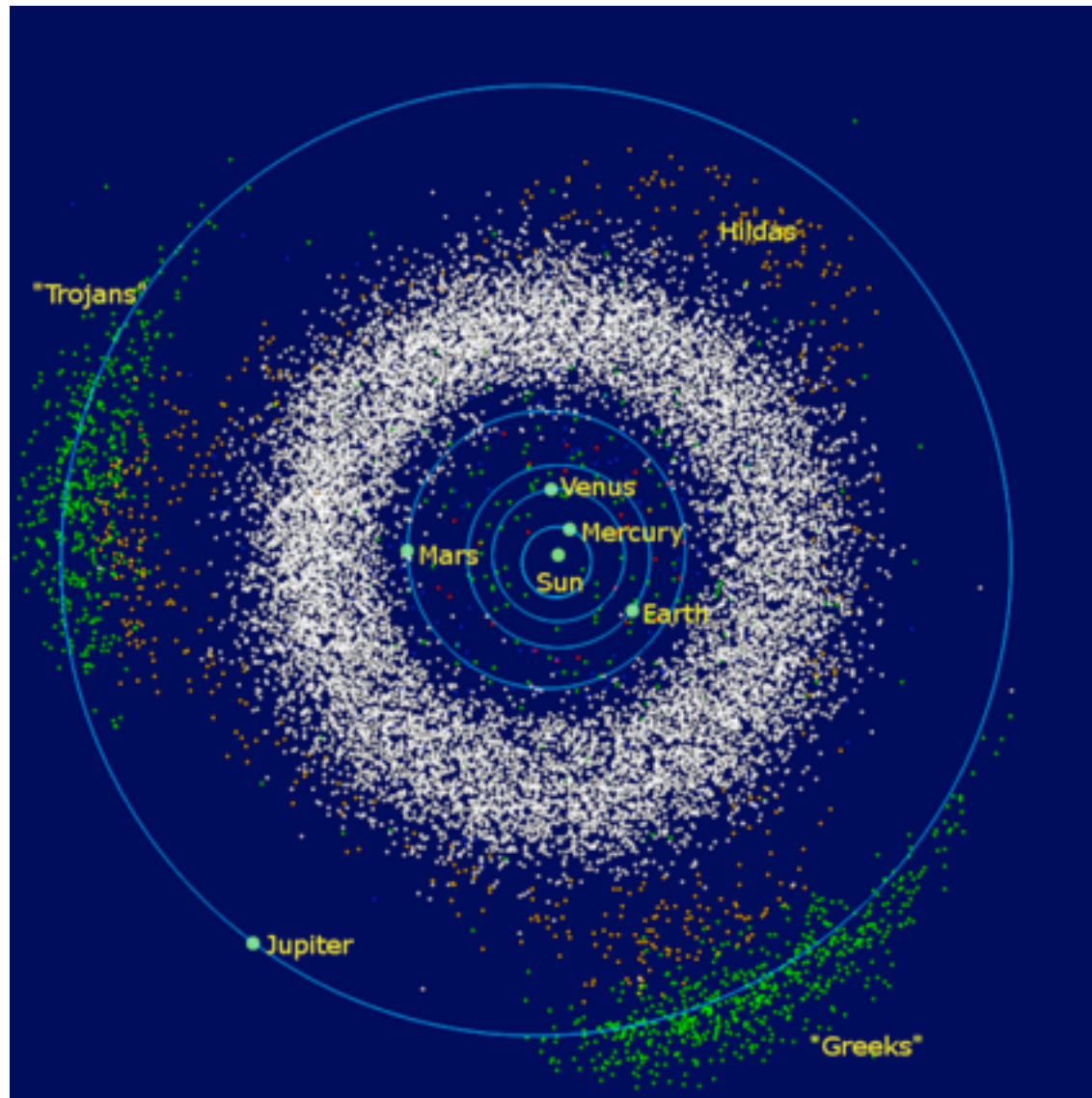
- ▶ Closer to the Sun
- ▶ Largely composed of rock/metal



## Comets

- ▶ Further from the Sun
- ▶ Largely made up of “icy” material

# Asteroids and the Asteroid Belt



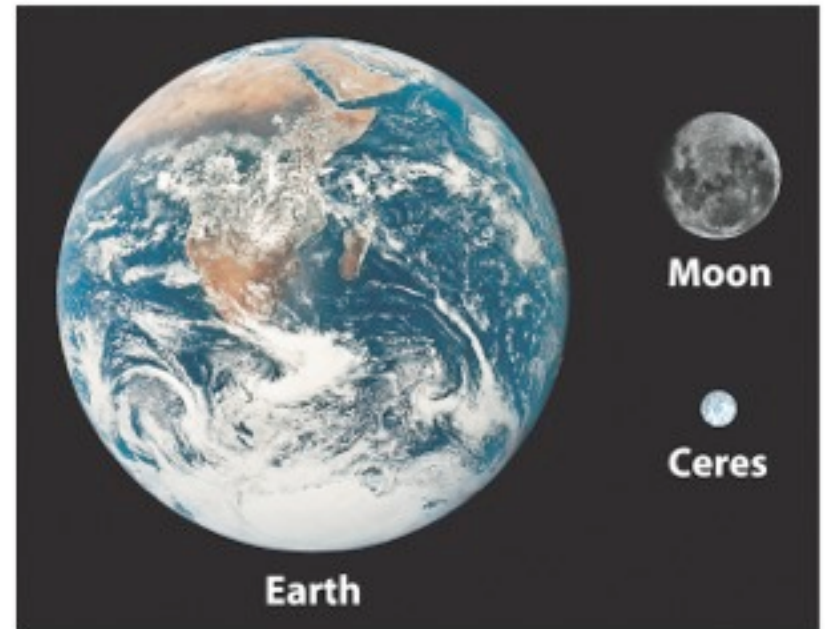
# Asteroids

## Small sizes

- ▶ Largest – **Ceres**: 940 km across
- ▶ Only 3 more than 300 km
- ▶ About 240 bigger than 100 km
- ▶ Millions under 1 km


## Composition

- ▶ **Rocks** (silicates)  
and **iron/nickel**



# Asteroids

M0151295144F4



December 3 2000 23:08:30 21° 146°

**Eros from NEAR:** <http://near.jhuapl.edu/iod/20010205/>

# What are asteroids like?

▶ Because they are small, they are **mostly unchanged since formation** 4.6 billion years ago

▶ Most have **irregular shapes: not spherical**

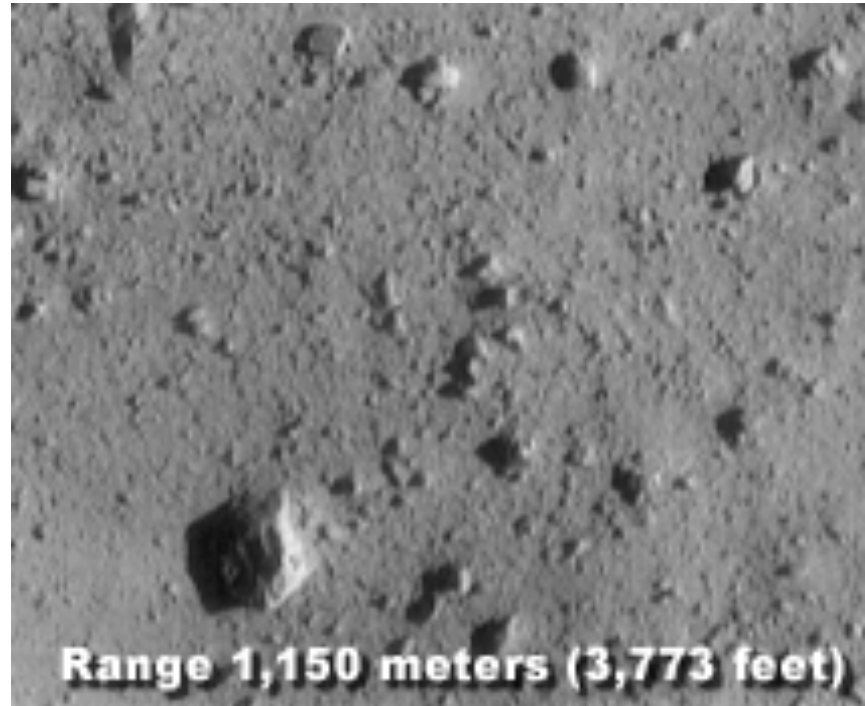
*Why not? Why are planets spherical?*

- Planets larger, more massive: gravity crushes them into sphere
- Asteroids small, low-mass, gravity too weak to overcome rock rigidity

▶ Pulverized rock “soil” like that of the Moon

▶ Boulders on surface

▶ Heavily cratered surfaces



**NEAR crash landing on asteroid Eros**

<http://www.youtube.com/watch?v=iiM7VHSRz4c>



# NEAR crash landing on asteroid Eros

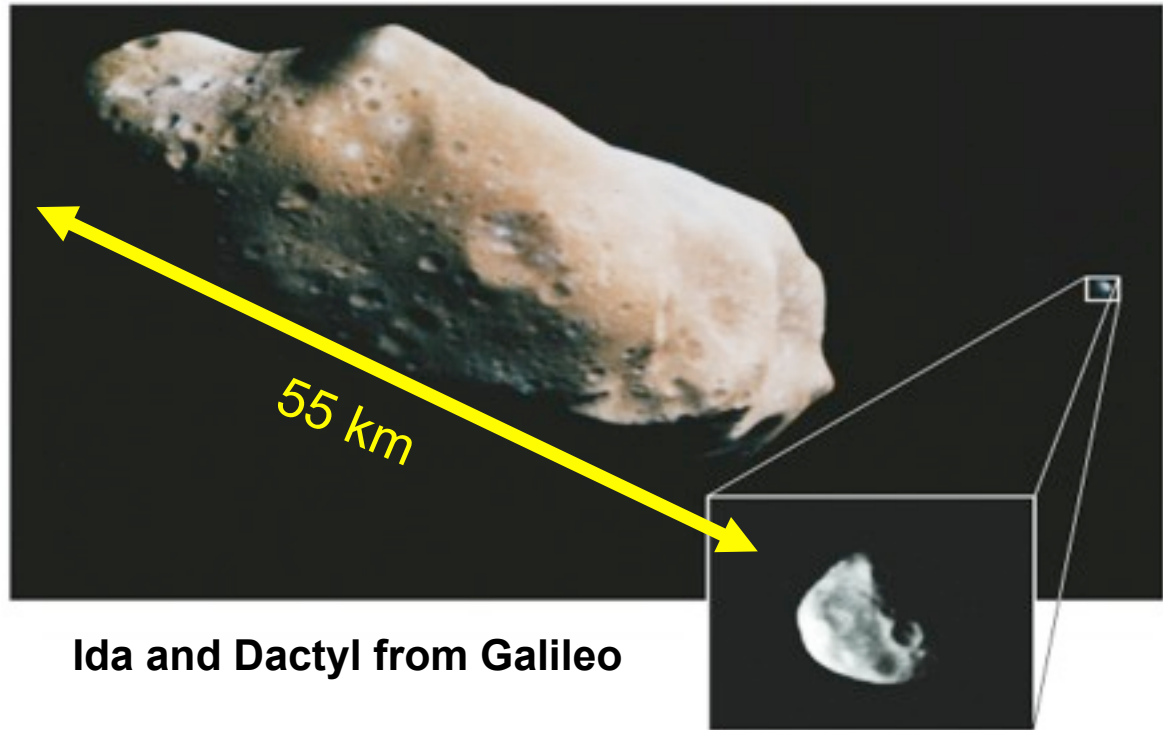


<http://www.youtube.com/watch?v=iiM7VHSRz4c>

# Asteroids with Moons

Some large  
asteroids have  
their own moon

Asteroid Ida has a  
tiny moon named  
Dactyl



# Rubble Pile Asteroids

- ▶ Rubble piles form when an asteroid is **smashed** to pieces by an **impact**
- ▶ Then the shattered pieces fall back together



**Asteroid 25143 Itokawa is a “rubble pile” asteroid**

Rubble piles have large cavities between the various 'chunks' that comprise them. Large interior voids are possible because of the very low gravity of most asteroids. Despite a fine regolith on the outside (at least to the resolution that has been seen with spacecraft), the asteroid's gravity is so weak that friction between fragments dominates and prevents small pieces from falling inwards and filling up the voids.